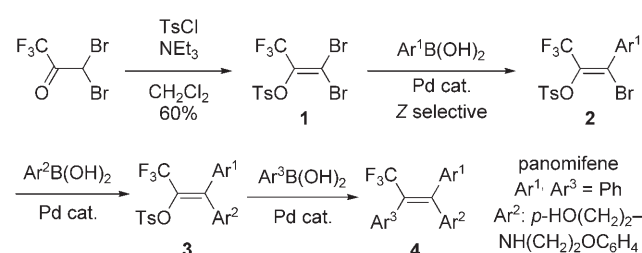


Cross-Coupling Reactions

Straightforward Synthesis of CF₃-Substituted Triarylethenes by Stereoselective Threefold Cross-Coupling Reactions**

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The incorporation of fluorine atoms into organic molecules has been a powerful strategy for the improvement of parent compounds in the fields of pharmaceuticals, agrochemicals, and organic materials.^[1] The preparation of panomifene^[2] is an example of the application of the strategy to triarylethenes, which constitute an important class of nonsteroidal anti-estrogens.^[3] Although a few stereoselective syntheses of CF₃-substituted triarylethenes are known, the availability of the CF₃-containing starting substrates and/or the use of organolithium or organomagnesium reagents limit their scope and versatility.^[4] We report herein that the threefold cross-coupling reaction of 1,1-dibromo-3,3,3-trifluoro-2-tosyloxypropene (**1**)^[5] with three kinds of aryl boronic acids provides a simple and straightforward approach to the synthesis of CF₃-substituted triarylethenes **4** (Scheme 1). The presence of a



Scheme 1. Approach for the synthesis of CF₃-substituted triarylethenes **4**. Ts = toluene-4-sulfonyl.

CF₃ group is essential for both high chemical yields and high *Z* selectivity in the first coupling reaction. This approach is highly versatile, since any stereoisomers of **4** can be arbitrarily synthesized in a stereochemically pure form simply by changing the order of the coupling partners, and, furthermore, a one-pot synthesis can also be achieved.

The twofold cross-coupling reaction of 1,1-dihalo-1-alkenes (RHC=CX₂) with organometallic compounds con-

stitutes an efficient method for the stereoselective synthesis of trisubstituted alkenes.^[6] Meanwhile, the threefold cross-coupling reaction of R(X²)C=C(X¹)₂ (X¹, X² = (pseudo)halogens) leading to unsymmetrical tetrasubstituted alkenes has no precedent, presumably because of the difficulty in discriminating between the three leaving groups (X¹ and X²) in the first coupling reaction.^[7] However, we anticipated that the steric and/or electronic nature of a CF₃ group might overcome this difficulty.^[8] Thus, we studied the palladium-catalyzed reaction of **1** with an equimolar amount of PhB(OH)₂ to prove the viability of the approach.^[9] Initial experiments disclosed that both the stereoselectivity of **2** and the production of decoupled and side products were critical issues to be solved.^[5] After extensive optimization, we found that monocoupled product **2a** could be isolated in 94% yield with high diastereoselectivity (92:8) when the reaction was conducted in the presence of [PdCl₂(PPh₃)₂] (5 mol %), P(*m*-tolyl)₃ (5 mol %), and 5 M aqueous Cs₂CO₃ in toluene at 80°C (Table 1, entry 1). The stereochemistry of the major product

Table 1: Coupling reaction of **1** with Ar¹B(OH)₂.^[a]

Entry	Ar ¹	2	Yield [%] ^[b]	<i>Z</i> / <i>E</i> ^[c]
1	C ₆ H ₅	2a	94	92:8
2	<i>p</i> -MeOC ₆ H ₄	2b	74	90:10
3	<i>p</i> -MeC ₆ H ₄	2c	73	89:11
4	<i>p</i> -CF ₃ C ₆ H ₄	2d	89	87:13
5	<i>p</i> -BrC ₆ H ₄	2e	85	88:12
6	<i>p</i> -PhC ₆ H ₄	2f	96	88:12
7	2-naphthyl	2g	78	89:11
8	<i>m</i> -FC ₆ H ₄	2h	87	89:11
9 ^[d]	3-thienyl	2i	80	90:10

[a] Reaction conditions: **2** (0.50 mmol), Ar¹B(OH)₂ (0.55 mmol), [PdCl₂(PPh₃)₂] (25 μmol), P(*m*-tolyl)₃ (25 μmol), 5 M aq Cs₂CO₃ (1.0 mmol), toluene (5 mL), 80°C, 24 h. [b] Yield of isolated product. [c] The *Z*/*E* ratio was determined by ¹⁹F NMR spectroscopic analysis of the crude products. [d] (3-Thienyl)B(OH)₂ (0.75 mmol) and 5 M aq K₃PO₄ (1.0 mmol) as a base were used.

was unambiguously confirmed as *Z* by single-crystal X-ray analysis,^[5] which indicated that a bromine atom *cis* to the CF₃ group was selectively substituted by a Ph group. Considering that the steric parameters (*A* values) of CF₃ and OTs groups are 2.4–2.5 and 0.5 kcal mol⁻¹, respectively,^[10] it should be noted that the coupling, in other words, an oxidative addition of **1** to a Pd center, occurs at the more hindered site. The optimized conditions were applicable to various aryl boronic acids, thus giving rise to **2b–i** in good to excellent yields (Table 1, entries 2–9). In the case of thienylboronic acid, the use of aqueous K₃PO₄ in place of aqueous Cs₂CO₃ resulted in

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awaits further study, a Pd...F interaction in the oxidative addition step may be operative.^[12]

In summary, we have developed a straightforward stereocontrolled synthesis of CF₃-substituted triarylethenes. Further study of the coupling reactions of 1,1-dibromo-3,3,3-trifluoro-2-tosyloxypropene with other kinds of nucleophiles is in progress.

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- [1] a) K. Uneyama, *Organofluorine Chemistry*, Blackwell, Oxford, **2006**; b) F. Leroux, P. Jeschke, M. Schlosser, *Chem. Rev.* **2005**, *105*, 827; c) R. D. Chambers, *Fluorine in Organic Chemistry*, Blackwell, Oxford, **2004**; d) P. Kirsch, *Modern Fluoroorganic Chemistry*, Wiley-VCH, Weinheim, **2004**; e) T. Hiyama, *Organofluorine Compounds. Chemistry and Applications*, Springer, Berlin, **2000**; f) *Houben-Weyl, Vol. E10a* (Eds.: B. Baasner, H. Hagemann, J. C. Tatlow), Thieme, Stuttgart, **2000**; g) *Biomedical Frontiers of Fluorine Chemistry ACS Symposium Series 639* (Eds.: I. Ojima, J. R. McCarthy, J. T. Welch), American Chemical Society, Washington, DC, **1996**; h) *Chemistry of Organic Fluorine Compounds II. A Critical Review ACS Monograph 187* (Eds.: M. Hudlicky, A. E. Pavlath), American Chemical Society, Washington, DC, **1995**; i) *Organofluorine Chemistry-Principles and Commercial Applications* (Eds.: R. E. Banks, B. E. Smart, J. C. Tatlow), Plenum, New York, **1994**; j) *Chemistry of Organic Fluorine Compounds*, 2nd ed. (Ed.: M. Hudlicky), Ellis Horwood, New York, **1992**; k) *Selective Fluorination in Organic and Bioorganic Chemistry ACS Symposium Series 456* (Ed.: J. T. Welch), American Chemical Society, Washington, DC, **1991**; l) *Fluorine-containing Molecules: Structure, Reactivity, Synthesis, and Applications* (Eds.: J. F. Liebman, A. Greenberg, J. W. R. Dolbier), VCH, New York, **1988**; m) J. T. Welch, *Tetrahedron* **1987**, *43*, 3123; n) M. Schlosser, *Tetrahedron* **1978**, *34*, 3; special issue on "Fluorine in the Life Sciences", o) *ChemBioChem* **2004**, *5*, 557–726.
- [2] a) J. Borvendeg, *Drugs Future* **1985**, *10*, 395; b) J. Borvendeg, I. Hermann, O. Csuka, *Acta Physiol. Hung.* **1996**, *84*, 405; c) V. Erdelyi-Toth, F. Gyergyay, I. Szamel, E. Pap, J. Kralovanszky, E. Bojtí, M. Csorgo, S. Drabant, I. Klebovich, *Anti-Cancer Drugs* **1997**, *8*, 603; d) K. Monostory, K. Jemnitz, L. Vereczkey, G. Czira, *Drug Metab. Dispos.* **1997**, *25*, 1370.
- [3] Reviews on antiestrogens: a) R. A. Magarian, L. B. Overacre, S. Singh, K. L. Meyer, *Curr. Med. Chem.* **1994**, *1*, 61; b) V. C. Jordan, *J. Med. Chem.* **2003**, *46*, 883; c) V. C. Jordan, *J. Med. Chem.* **2003**, *46*, 1081.
- [4] a) W. J. Middleton, D. Metzger, J. A. Snyder, *J. Med. Chem.* **1971**, *14*, 1193; b) G. Németh, R. Kapiller-Dezsofi, G. Lax, G. Simig, *Tetrahedron* **1996**, *52*, 12821; c) X. Liu, M. Shimizu, T. Hiyama, *Angew. Chem.* **2004**, *116*, 897; *Angew. Chem. Int. Ed.* **2004**, *43*, 879; d) T. Konno, T. Daitoh, A. Noiri, J. Chae, T. Ishihara, H. Yamanaka, *Org. Lett.* **2004**, *6*, 933; e) M. S. Kim, I. H. Jeong, *Tetrahedron Lett.* **2005**, *46*, 3545; f) T. Konno, T. Daitoh, A. Noiri, J. Chae, T. Ishihara, H. Yamanaka, *Tetrahedron* **2005**, *61*, 9391.
- [5] See the Supporting Information.
- [6] a) T. Sugihara in *Handbook of Organopalladium Chemistry for Organic Synthesis*, Vol. 1 (Ed.: E. Negishi), Wiley, New York, **2002**, 649; b) O. Reiser, *Angew. Chem.* **2006**, *118*, 2904; *Angew. Chem. Int. Ed.* **2006**, *45*, 2838; c) A. Minato, K. Suzuki, K. Tamao, *J. Am. Chem. Soc.* **1987**, *109*, 1257; d) W. R. Roush, K. J. Moriarty, B. B. Brown, *Tetrahedron Lett.* **1990**, *31*, 6509; e) X. Zeng, M. Qian, Q. Hu, E. I. Negishi, *Angew. Chem.* **2004**, *116*, 2309; *Angew. Chem. Int. Ed.* **2004**, *43*, 2259; f) G. A. Molander, Y. Yokoyama, *J. Org. Chem.* **2006**, *71*, 2493; g) Z. Tan, E. I. Negishi, *Angew. Chem.* **2006**, *118*, 776; *Angew. Chem. Int. Ed.* **2006**, *45*, 762.
- [7] Palladium-catalyzed reactions of RR'C=CBr₂ to give unsymmetrical tri- and tetrasubstituted alkenes have been reported; hydrogenolysis: a) J. Uenishi, R. Kawahama, O. Yonemitsu, J. Tsuji, *J. Org. Chem.* **1998**, *63*, 8965; intramolecular bis(carbopalladation): b) S. Ma, B. Xu, B. Ni, *J. Org. Chem.* **2000**, *65*, 8532; cross-coupling reactions with organostannanes: c) A. Sorg, K. Siegel, R. Brückner, *Synlett* **2004**, 321; see also, d) R. C. Larock, M. J. Doty, X. Han, *J. Org. Chem.* **1999**, *64*, 8770; the sequential synthesis of tetraaryl ethenes starting from vinyl sulfide has been reported: e) K. Itami, M. Mineno, N. Muraoka, J. I. Yoshida, *J. Am. Chem. Soc.* **2004**, *126*, 11778.
- [8] Bromine–lithium and bromine–zinc exchange in 2-substituted 1,1-dibromo-3,3,3-trifluoropropenes (R(CF₃)C=CBr₂; R = carbonaceous group) were reported to take place stereoselectively: a) A. Morken, P. C. Bachand, D. C. Swenson, D. J. Burton, *J. Am. Chem. Soc.* **1993**, *115*, 5430; b) H. Uno, N. Nibu, N. Misobe, *Bull. Chem. Soc. Jpn.* **1999**, *72*, 1365; c) Y. Li, L. Lu, X. Zhao, *Org. Lett.* **2004**, *6*, 4467.
- [9] Reviews on Suzuki–Miyaura coupling: a) N. Miyaura, A. Suzuki, *Chem. Rev.* **1995**, *95*, 2457; b) A. Suzuki, in *Metal-catalyzed Cross-coupling Reactions* (Eds.: F. Diederich, P. J. Stang), Wiley-VCH, Weinheim, **1998**, p. 49; c) N. Miyaura, *Top. Curr. Chem.* **2002**, *219*, 11; d) N. Miyaura, in *Metal-Catalyzed Cross-Coupling Reactions, Vol. 1* (Eds.: A. de Meijere, F. Diederich), Wiley-VCH, Weinheim, **2004**, p. 41.
- [10] E. L. Eliel, S. H. Wilen, L. N. Mander, *Stereochemistry of Organic Compounds*, Wiley, New York, **1994**, p. 690–700.
- [11] H. N. Nguyen, X. Huang, S. L. Buchwald, *J. Am. Chem. Soc.* **2003**, *125*, 11818.
- [12] C. Bartolome, P. Espinet, F. Villafane, S. Giesa, A. Martin, A. G. Orpen, *Organometallics* **1996**, *15*, 2019.